

Chapter 9 Stoichiometry Section 2 Worksheet

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Chapter 9 Stoichiometry Section 2

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SECTION 9.2 Ideal Stoichiometric Calculations

SECTION 92 Balanced equations give amounts of reactants and Stoichiometry 287 SAMPLE PROBLEM In a spacecraft, the carbon dioxide exhaled by astronauts can be removed by its reaction with lithium hydroxide, LiOH, according to Refer to Section 2 of the chapter "Chemical Equations and

CHAPTER 9 REVIEW Stoichiometry

Modern Chemistry 75 Stoichiometry CHAPTER 9 REVIEW Stoichiometry SECTION 2 PROBLEMS Write the answer on the line to the left Show all your work in the space provided 1 ____ The following equation represents a laboratory preparation for oxygen gas: $2\text{KClO}_3(\text{s}) \rightarrow 2\text{KCl}(\text{s}) + 3\text{O}_2(\text{g})$ How many moles of O_2 form if 30 mol of KClO_3 are

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factors Key Terms composition stoichiometry

Chapter 9 Review Stoichiometry Section 2 Answers

Chapter 9 Stoichiometry Section 2 Worksheet SECTION 2 continued Date Class ____ 602 9 421 1 a \ tt mash 01 ox aen Cas i priduied it 100 of lithium c a C ti I o c i o g di I C1O c — LCi(,; — h The oxygen gas produced in part ahas density of 143 g/L aiculate the olurne of

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Chapter 9 Stoichiometry

Chapter 9 Section 1 Introduction to Stoichiometry Lesson Starter $Mg(s) + 2HCl(aq) \rightarrow MgCl_2(aq) + H_2(g)$ • If 2 mol of HCl react, how many moles of H₂ are obtained? 1 mol H₂ • How many moles of Mg will react with 2 mol of HCl?

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SECTION 2 continued Date Class ____ 602 9 421 1 a \ tt mash 01 ox aen Cas i priduied it 100 of lithium c a C ti I o c i o g di I C1O c — L Ci(,; — h The
 oxygen gas produced in part a has density of 1.43 g/L calculate the volume of this gas 76 STOICHIOMETRY MODERN CHEMISTRY a — 81 g 6 A car air
 bag requires 70 L of nitrogen gas

Chapter 9 - Stoichiometry

9-2 Ideal Stoichiometric Calculations Ideal Stoichiometry - All reactants are converted into products I A Common Method for Solving All
 Stoichiometry Problems A Mass-Mass Problems 1 Start with a known mass of reactant or product, find an unknown mass of another reactant or
 product 2 All other stoichiometry problems are derivations